

$$w = \int P \, dv \quad \text{Quasi-equilibrium (reversible) process;} \quad h = u + Pv \quad \text{Definition, always true}$$

$$PV = mRT = n\bar{R}T \quad \text{Ideal gas (} T > 2^*T_{\text{crit}}, P < 10 \text{ MPa)} \quad u \approx u_f @ T; \quad v \approx v_f @ T \quad \text{Subcooled liquid}$$

$$Q - W = E_2 - E_1 \quad \text{First law for a process, no constraints}$$

$$Q - W = U_2 - U_1 \quad \text{First law for a process, closed system, negligible changes in kinetic and potential energy}$$

$$C_v = \frac{\partial u}{\partial T}, \quad C_p = \frac{\partial h}{\partial T} \quad \text{Definition;} \quad C_v = \frac{du}{dT}, \quad C_p = \frac{dh}{dT} \quad u \text{ and } h \text{ functions of temperature only}$$

$$u_2 - u_1 = \int C_v \, dT \quad \text{Internal energy is a function of temp only;} \quad h_2 - h_1 = \int C_p \, dT \quad \text{Enthalpy is a function of temp only}$$

$$\oint \delta Q = \oint \delta W \quad \text{Any thermodynamic cycle} \quad \dot{m} = \rho \vec{V} A \quad \text{steady, 1-d flow}$$

$$\dot{Q} + \sum \dot{m}_i \left(h + \frac{\vec{v}^2}{2} + gZ \right)_i = \dot{W} + \sum \dot{m}_e \left(h + \frac{\vec{v}^2}{2} + gZ \right)_e + \frac{dE}{dt} \quad \text{First law for control volume}$$

$$q + h_1 + \frac{1}{2}\vec{V}_1^2 + gZ_1 = w + h_2 + \frac{1}{2}\vec{V}_2^2 + gZ_2 \quad \text{Single-inlet, single-exit, steady flow}$$

$$\eta_{\text{th}} = \frac{W_{\text{net}}}{Q_H} = 1 - \frac{Q_L}{Q_H} \quad \text{Heat engine;} \quad \beta = \frac{Q_L}{W_{\text{net}}} \quad \text{Heat pump (cooler);} \quad \gamma = \frac{Q_H}{W_{\text{net}}} \quad \text{Heat pump (heater)}$$

$$w = - \int v \, dP - \frac{1}{2} (\vec{V}_2^2 - \vec{V}_1^2) - g(Z_2 - Z_1) \quad \text{Reversible, steady flow}$$

$$q = \int T \, ds \quad \text{Reversible process}$$

$$Pv^n = \text{constant} \quad \text{Polytropic process (ideal gas) (} n = 0, P = c; \quad n = 1, T = c; \quad n = k, s = c; \quad n \rightarrow \infty, v = c)$$

$$\frac{T_2}{T_1} = \left(\frac{P_2}{P_1} \right)^{\frac{k-1}{k}}; \quad \frac{T_2}{T_1} = \left(\frac{v_2}{v_1} \right)^{1-k}; \quad \frac{P_2}{P_1} = \left(\frac{v_1}{v_2} \right)^k \quad \text{Isentropic, ideal gas, constant specific heats}$$

$$\Delta s = C_p \ln \frac{T_2}{T_1} - R \ln \frac{P_2}{P_1}, \quad \Delta s = C_v \ln \frac{T_2}{T_1} + R \ln \frac{v_2}{v_1} \quad \text{Ideal gas, constant specific heats}$$

$$\frac{P_{R2}}{P_{R1}} = \frac{P_2}{P_1}; \quad \frac{v_{R2}}{v_{R1}} = \frac{v_2}{v_1} \quad \text{Isentropic, ideal gas, non-constant specific heats}$$

$$\eta_{\text{turbine}} = w_a/w_s \quad \eta_{\text{compressor}} = w_s/w_a \quad \eta_{\text{pump}} = w_s/w_a \quad \eta_{\text{nozzle}} = \frac{\vec{V}_a^2}{\vec{V}_s^2}$$

